

a) Isothermer Prozess

$$p_1 V_1 = p_0 V_0$$

$$p_1 A h_1 = p_0 A h_0 \quad \Rightarrow \quad h_1 = \frac{p_0 h_0}{p_1}$$

$$h_1 = \frac{p_0 h_0}{p_0 + \rho_{\text{Hg}} \cdot g \cdot l} = \frac{980 \text{ hPa} \cdot 50 \text{ cm}}{980 \text{ hPa} + 13,55 \text{ g/cm}^3 \cdot 9,81 \text{ m/s}^2 \cdot 0,1 \text{ m}}$$

$$= \underline{\underline{44 \text{ cm}}}$$

b) isobar $\frac{V_2}{T_2} = \frac{V_1}{T_1} \quad \Rightarrow \quad \frac{A h_2}{T_2} = \frac{A h_1}{T_1}$

$$\Rightarrow h_2 = \frac{373}{295} \cdot 44 \text{ cm} = \underline{\underline{56 \text{ cm}}}$$

$$c) \left(p_2 + \frac{a m^2}{V_2} \right) (V_2 - b m) \cdot \frac{1}{T_2} = \left(p_1 + \frac{a m^2}{V_1} \right) (V_1 - b m) \cdot \frac{1}{T_1}$$

$$(V_2^2 p_2 + a m^2) (V_2 - b m) \frac{1}{T_2} = (V_2^2 p_1 + a m^2 \frac{V_2^2}{V_1^2}) (V_1 - b m) \frac{1}{T_1}$$

$$V_2^3 p_2 + V_2^2 p_2 b m + a m^2 V_2^2 - a b m^3 =$$

$$\frac{T_2}{T_1} \left[V_2^2 p_1 \cdot V_1 - V_2^2 p_1 b m + a m^2 \frac{V_2^2}{V_1} - a b m^3 \frac{V_2^2}{V_1^2} \right]$$

$$V_2^3 p_2 + V_2^2 \left(-p_2 b m - \frac{T_2}{T_1} p_1 V_1 + \frac{T_2}{T_1} p_1 b m - \frac{T_2}{T_1} a m^2 \frac{1}{V_1} + \frac{T_2}{T_1} a b m^3 \frac{1}{V_1^2} \right)$$

$$+ V_2 \cdot a m^2 - a b m^3 = 0$$

→ Ausrechnen sehr lästig!

